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SHORT COMMUNICATIONS

Reaction of 5-Nitrobenzofuroxane with Acetylacetone^{*}

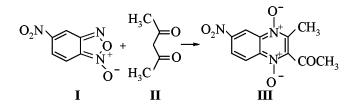
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of 5(6)-substituted benzofuroxanes Reactions with C-nucleophiles (enolate-anions, enamines) resulting in quinoxaline-N,N'-dioxides frequently provide better yields than those with unsubstituted benzofuroxane [1, 2]. However the substituents in 4(7) position sharply decrease the reactivity although at long time and elevated temperature the condensation product can be obtained in low yield [2]. At the same time both 4- and 6-nitrobenzofuroxanes vigorously react with enamines [2] to afford a complicated mixture of products that does not contain quinoxaline-N, N'-dioxides. Similar results were also obtained at low temperature [2].

Here we report on successful condensation of 5-nitrobenzofuroxane (I) with acetylacetone (II) in the presence of calcium hydroxide or alcoholate.



The structure of obtained 2-acetyl-3-methyl-6nitroquinoxaline-N,N'-dioxide was confirmed by ¹H NMR and IR spectra. The position of the substituents was determined basing on published data [3, 4] indicating that at an electron-withdrawing substituent in 5(6) position formed prevailingly type **III** product. The reaction time depends on benzofuroxane to catalyst ratio; at equimolar amounts it is ~2 h, yield 65-67% (catalyst calcium isopropylate), and 50-55% [catalyst Ca(OH)₂].

2-Acetyl-3-methyl-6-nitroquinoxaline-*N*, *N*'**dioxide.** A solution of 1.81 g (10 mmol) of 5-nitrobenzofuroxane, 1.1 g (11 mmol) of acetylacetone in 50 ml of 2-propanol was mixed with a solution of 0.25 g (2.4 mmol) of calcium isopropylate in 10 ml of 2-propanol. The mixture was kept for 5 h at 18-20°C, The separated precipitate was filtered off and recrystallized from acetone. We obtained 1.75 g (67%) of orange needle-like crystals, mp 162°C (decomp.). IR spectrum, v, cm⁻¹: 1720, 1620, 1537, 1420, 1335, 1250, 1180, 1050, 920. ¹H NMR spectrum (CF₃COOH), δ , ppm: 2.35 s (3H, COCH₃), 2.4 s (3H, CH₃), 8.4 q (2H, H^{7.8}), 9.15 s (1H, H⁵). Found, % : C 50.16; H 3.51; N 15.89. C₁₁H₉N₃O₅. Calculated,%: C 50.20; H 3.45; N 15.96.

¹H NMR spectra were registered om spectrometer Perkin-Elmer R-12, internal reference HMDS. TLC was performed on Silufol UV-254 plates, eluent chloroform.

REFERENCES

- 1. Issidorides, C.H., Agopian, G., and Haddadin, M.J., *J. Org. Chem.*, 1971, vol. 36, no. 4, pp. 514–517.
- Mufarrij, N.A., Haddadin, M.J., and Issidorides, C.H., J. Chem. Soc., Perkin Trans. I, 1972, pp. 965–967.
- 3. Abushanab, E. and Alteri, N.D., J. Org. Chem., 1975, vol. 40, no. 2, pp. 157-160.
- 4. Durckheimer, W., *Lieb. Ann.*, 1972, vol. 756, pp. 145–154.

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